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Biodegradation of toluene by the new fungal isolates *Paecilomyces variotii* and *Exophiala oligosperma*

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Abstract Two new fungal strains, namely Paecilomyces variotii and Exophiala oligosperma, were isolated on toluene as the sole carbon and energy source, mineralizing the substrate into carbon dioxide. Fungal strains isolated so far on such a pollutant and completely degrading it are very scarce. Both fungi degraded the pollutant over the pH range 3.9-6.9 and temperature range 23-40°C, but E. oligosperma was barely active at the highest temperature of 40°C. Fungal growth on alkylbenzenes at 40°C has not been reported before. Since the activity of the strains gradually decreased at pH values below 4.0, the use of nitrate instead of ammonium was tested. In the presence of toluene, nitrate was a suitable nitrogen source for the Exophiala strain, but not for the Paecilomyces strain. Nitrate rather than ammonium allowed the maintenance of a more constant pH.

Keywords Alkylbenzenes · Biofiltration · Bioremediation · Extremophiles · Volatile organic compounds

Introduction

Alkylbenzenes are common contaminants in soils, aquifers and waste gases. Together with benzene, they are among the most abundant aromatic compounds in gasoline and some other refined products obtained from crude oil. In addition, they are used in many industrial processes as solvents and they provide the starting material for the production of many other chemicals. Benzene and alkylbenzenes released into the environ-

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ment are present at low concentrations (in the parts per million range) and are degradable by different bacterial species [15, 16].

The use of alkylbenzenes as sole carbon and energy sources in bacteria, which was demonstrated several decades ago, has been studied extensively in such organisms [1]. However, although the biodegradation of such pollutants is well documented in prokaryotes, a similar characteristic was not demonstrated in filamentous fungi until the mid-1990s, when a Cladosporium sphaerospermum strain was described for its ability to grow on toluene as sole carbon and energy source while completely mineralizing it [19]. As reviewed recently, only very few other fungal strains have been identified and isolated since then on alkylbenzenes [10]. White-rot fungi are also known to degrade volatile aromatic pollutants, although this is often a co-metabolic process in those organisms. Besides, the biodegradation of toluene in different strains of the white-rot fungus *Phanerochaete* chrysosporium led to only about 50% mineralization in the best case [21].

The recently reported complete biodegradation of some aromatic volatile organic compounds (VOC) by a few filamentous fungi is an interesting discovery, not only for the progress of basic research but also for its potential application to the treatment of VOC-contaminated systems, in particular waste gases [9, 10]. Indeed, it was recently shown that higher toluene removal efficiencies can be reached in biofilters in which filamentous fungi rather than bacteria are dominant [4, 5, 9, 18, 20]. It is hypothesized that the filamentous structure of these eukaryotes allows for a rapid mass transfer of hydrophobic pollutants from the gas phase to the biocatalyst [6, 9]. Besides, fungi are more resistant than most bacteria to extreme environmental conditions such as low pH, reduced water content, etc. [5, 9, 10]. Such characteristics are also interesting in soil bioremediation in environments hostile to bacterial activity [14].

In this paper, the isolation on toluene of two new fungal strains, *Paecilomyces variotii* and *Exophiala oligosperma*, with complete mineralization of the substrate, is reported. Their activity was evaluated at different temperatures and pH and with two different nitrogen sources. This is the first report on the biodegradation of this alkylbenzene by these fungal species and on the consumption of toluene at low pH and relatively high temperature (40° C) in fungi.

Materials and methods

Media composition

Batch experiments were undertaken with an aqueous culture medium containing (per liter): 4.5 g KH₂PO₄, 0.5 g K₂HPO₄, 2.0 g NH₄Cl and 0.1 g MgSO₄·H₂O. The culture medium was autoclaved at 120°C for 20 min before adding filter-sterilized solutions of vitamins and trace minerals. The composition of the vitamins solution was (per liter): 0.2 g thiamine HCl, 0.1 g riboflavin, 1.0 g nicotinic acid, 2.0 g Ca-pantothenate, 0.1 g biotin, 0.1 g thioctic acid, 0.1 g folic acid and 0.25 g pyridoxine HCl. The composition of the trace minerals solution was (per liter): 120 mg FeCl₃, 50 mg H₃BO₃, 10 mg CuSO₄·5- H_2O , 10 mg KI, 45 mg MnSO₄· H_2O , 20 mg Na₂-MoO₄·2H₂O, 75 mg ZnSO₄·7H₂O, 50 mg CoCl₂·6H₂O, 20 mg AlK(SO₄)₂·12H₂O, 13.25 mg CaCl₂·2H₂O and 10,000 mg NaCl. The original pH of that medium was 5.9. The pH could be adjusted to another value either by modifying the mass ratio of the phosphate compounds while maintaining a constant concentration of potassium or, most often, by adding NaOH or HCl. When studying the effect of the nitrogen source, NH₄Cl was replaced by KNO₃.

Stock cultures of the fungi were maintained on Petri dishes or on slants using either potato dextrose agar (PDA) or the same mineral medium as described above supplemented with 16 g agar 1^{-1} . When using the mineral medium, the plates were incubated in a tank or desiccator at 30°C, in the presence of toluene vapors as carbon source. Stock cultures on PDA were stored in a refrigerator at 4°C.

Alkylbenzene biodegradation in batch assays

All assays were performed at least in duplicate, with the corresponding controls. Both uninoculated media (called "blanks") and inoculated autoclaved vials (called "controls") were used. The mineral medium described above (200 ml) was introduced into 500-ml bottles closed with Viton septa and screw caps. The pollutant was added, to reach a gas-phase concentration of about 6 mg 1^{-1} air, unless otherwise stated. Although data are reported as gas phase concentrations, the substrate is a volatile compound and is distributed between the gas and liquid phases. As described below, biodegradation was followed by sampling the gas phase. The concentration in the liquid phase can easily be calculated based on Henry's partition coefficient. Stock cultures of the

desired fungus were used as inoculum. Stock cultures were grown in mineral medium with toluene until they reached an optical density of 0.3 at 660 nm. After homogenization, 10 ml of that stock culture were inoculated into the bottles for the biodegradation assays, allowing all experiments to start with identical biomass concentrations and allowing the calculation of specific biodegradation rates (micrograms of toluene per gram of biomass per minute). The bottles were maintained in a thermostatic shaker at 30°C, unless otherwise stated, with constant shaking at 200 rpm.

Analytical methods

Air samples were withdrawn from the vials under sterile conditions by means of a gas-tight syringe. The samples were then analyzed on a HP6890 gas chromatograph (GC) equipped with a flame ionization detector and a TR-Wax tracer column according to a method described elsewhere [18]. Carbon dioxide was measured either by a gas sensor or by gas chromatography on a HP6890 GC equipped with a thermal conductivity detector and a Porapack Q column W80/100. Samples for SEM photographs were prepared according to a previously published procedure [12].

Results and discussion

Origin of the fungi and their substrate range

The two new fungi were isolated from a biofilter that was treating alkylbenzene-polluted air. The biofilter was originally inoculated with a defined consortium described elsewhere [17], composed of a *Bacillus* strain, a Pseudomonas strain and the yeast Trichosporon beigelii. Its maximum elimination capacity was around 70 g m⁻³ h⁻¹. However, after more than 1 year of operation, heavy fungal contamination was observed and, around that same period, the elimination capacity increased to 120 g m⁻³ h⁻¹, with 100% removal efficiency [18]. Two fungi were isolated on toluene at the end of the second year of operation, from the highest dilutions of samples of the carrier material serially diluted in a physiological solution, composed of 0.09% NaCl in distilled water. They were identified at the Centraal Bureau voor Schimmelcultures (The Netherlands) as E. oligosperma and Paecilomyces variotii, and were deposited as CBS-113408 and CBS-113409, respectively. Each organism formed a filamentous network when grown in packed-bed reactors such as gas-phase biofilters (Fig. 1a,b). This is an interesting characteristic, since it has been suggested that the growth of filamentous organisms enhances the mass transfer of hydrophobic pollutants from the gas (air) to the biocatalyst. E. oligosperma is a new, recently described, fungal species [7]. Each strain was transferred to the aqueous nutritive solution at pH 5.9 with toluene as



Fig. 1 SEM pictures of filter-bed samples from gas-phase packedbed bioreactors colonized by cultures of either **a** *E. oligosperma* or **b** *P. variotii*

the sole carbon source, demonstrating their ability to use that compound as a carbon and energy source (Fig. 2). No soluble intermediate product was detected in the medium throughout the course of the biodegradation study nor after complete exhaustion of the substrate. Carbon dioxide was identified as the end-product, with an average recovery of about 65%, the rest of the sub-



Fig. 2 Biodegradation of toluene by *E. oligosperma* (\times) and *P. variotii* (+) at pH 5.9. Data for one control (*filled squares*) and one blank (*open squares*) are also plotted

strate being converted into biomass. The specific toluene biodegradation rate reached $37 \pm 3 \ \mu g_{toluene} \ g^{-1} \ min^{-1}$ in the case of *E. oligosperma*, while it was only $16 \pm 2 \ \mu g_{toluene} \ g^{-1} \ min^{-1}$ with *P. variotii*.

The same procedure was followed in order to check the eventual biodegradation of other benzene compounds as single substrates by the isolated fungi. The following carbon sources were tested with both isolates: benzene, ethylbenzene, o-, m-, p-xylene and styrene. None of these compounds was biodegraded. The narrow substrate range of fungi able to grow on toluene seems to be a common feature of the few species recently described in the literature [10]. So far, only one fungal strain, Cladophialophora sp. strain T1, has been shown to be able to degrade toluene, ethylbenzene and xylenes, although the xylenes were only degraded co-metabolically and were not consumed when supplied as a singlealkylbenzene substrate [13]. Although P. variotii forms pellets when grown in aqueous phase in shake-flasks, recent studies show that each fungus produces large filaments when grown on a solid support in gas-phase biofilters [2, 3] (Fig. 1a,b). Our on-going biofiltration studies with E. oligosperma indicate that higher elimination capacities can be reached with that fungal isolate than in conventional bacteria-dominant biofilters, suggesting the advantage of using fungi-dominant biofilters. Indeed, preliminary data show that the elimination capacity can reach values as high as 164 g m⁻³ h⁻¹ with that organism [3], while values ranging between approximately $10 \text{ g m}^{-3} \text{ h}^{-1}$ and $70 \text{ g m}^{-3} \text{ h}^{-1}$ are typical of conventional bacterial biofilters [8, 9].

Effect of temperature on the biodegradation of toluene

Only a few fungi can grow under thermophilic conditions [11] and none has been reported to grow on toluene at higher temperatures than about 30°C. Most previous studies with alkylbenzenes were performed between room temperature (i.e. about 20°C) and 30°C. Since few *Paecilomyces* strains growing on sugars are known to be thermophilic [11], it was decided to compare the rate of toluene biodegradation at temperatures between 23°C and 40°C at pH 5.9. As shown in Fig. 3, *P. variotii* was



Fig. 3 Effect of temperature on the biodegradation activity of *P. variotii* (+) and *E. oligosperma* (×). The highest activity reached with *E. oligosperma*, as mentioned in the text, is used as a reference and is normalized to 100% to allow easy comparison between the different strains and conditions



Fig. 4 Effect of pH on the biodegradation activity of *P. variotii* (+) and *E. oligosperma* (×). The highest activity reached with *E. oligosperma*, as mentioned in the text, is used as areference and is normalized to 100% to allow easy comparison between the different strains and conditions



Fig. 5 Biodegradation of toluene by *E. oligosperma* (\times) and *P. variotii* (+) at pH 5.9 with nitrate as single nitrogen source. Data for one control (*open squares*) and one blank (*filled squares*) are also plotted

able to grow on toluene over the temperature range tested. Conversely, in the case of *E. oligosperma*, the rate of toluene biodegradation was relatively constant between 23°C and 35°C and was higher than with *P.variotii*, but dropped significantly at 40°C (Fig. 3). These data suggest the possibility of using fungal bio-filters under slightly thermophilic conditions. The isolation of new thermophilic species will broaden the field of application of fungal biofilters to a wider number of pollution sources.

Effect of pH on the biodegradation of toluene

Fungi are usually relatively tolerant to acidic conditions, compared with bacteria. A set of experiments was undertaken with both fungal strains in buffered media at different pH values between 3.9 and 6.9. The temperature was maintained at 30°C. In this case, both fungi degraded toluene over the whole pH range tested (Fig. 4). The optimum pH was close to 5.9. Although inhibition was somewhat more significant at the lowest pH for *E. oligosperma* than for the *Paecilomyces* strain, its biodegradation rate was higher at pH 3.9 than that of *P. variotii* since, as mentioned above, the former strain degrades toluene at a higher rate than the latter under optimal conditions. Acidification of filter beds has sometimes been observed in biofilters used for air pol-

lution control; and thus the development of microorganisms active under acidic conditions, such as the present fungal isolates, is a prerequisite for the successful operation of such systems.

Toluene degradation with different nitrogen sources

Although both fungi can grow efficiently on toluene at low pH, the biodegradation rates gradually decrease with lower pH values. As shown in the previous experiments, both fungi can use ammonium as their nitrogen source. Since the consumption of ammonium leads to the release of protons, the pH may drop in poorly buffered media or, for example, in gas-phase biofilters. This is a major concern in cases where acidification may negatively affect the performance of the reactors. Therefore, the ability to simultaneously degrade toluene and use nitrate was checked in both strains, since acidification is much less with such a nitrogen source. The results indicate that E. oligosperma was able to use nitrate, although P. variotii was not (Fig. 5). Although the biodegradation rate of toluene in the presence of nitrate was slightly higher than with ammonium (Fig. 2), the difference was not significant.

Conclusions

- Two new fungal strains, *E. oligosperma* and *P. variotii*, were shown to use toluene as their sole carbon and energy source and to mineralize it to carbon dioxide.
- *P. variotii* degraded toluene at 23–40°C, while *E. oligosperma* was inactive at the highest temperature of 40°C.
- Both strains degraded toluene at pH 3.9–6.9, with an optimal activity around pH 5.9.
- E. oligosperma was able to use either ammonium or nitrate as nitrogen source in the presence of toluene, but *P. variotii* used only ammonium.

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